

“Die Bewegungsgleichungen des Elektrons und Prinzip der kleinsten Aktion.” Sitz. Kais. Akad. Wiss. in Wien, math.-naturw. Klasse, Abt. 2.a, **120** (1911), 161-164.

The equations of motion of the electron and the principle of least action

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As is known, **Hölder** and **Voss** ⁽¹⁾ have presented the principle of least action in the general form:

$$\int_{t_0}^{t_1} [\delta L \cdot dt + 2L \cdot d\delta t + \delta U' \cdot dt] = 0 ,$$

in which L represents the present energy and $\delta U'$ is the elementary external force, and time is also varied. That equation is true for *all purely-mechanical* processes, but breaks down as soon as the principle is applied to reversible, *not purely-mechanical*, processes, e.g., to thermodynamics, electrodynamics, etc. It would then be advisable to write the principle in the form:

$$\int_{t_0}^{t_1} [\delta H \cdot dt + (H + E) \cdot d\delta t + \delta U \cdot dt] = 0 , \quad (\text{I})$$

in which H is now the *kinetic potential*, which is a function of the general coordinates p_i and \dot{p}_i , $\delta U = \sum P_i \delta p_i$ represents the elementary external work, and the quantity E , which later proves to be the *energy*, is defined by the relation:

$$E = \sum \dot{p}_i \frac{\partial H}{\partial \dot{p}_i} - H . \quad (\text{II})$$

Namely, if one imagines that in our case the variation $\delta\dot{\psi}$ of a velocity $\dot{\psi}$ obeys the rule:

⁽¹⁾ Göttinger Nachrichten, 1896 and 1900.

$$\delta\dot{\psi} = \frac{d\delta\psi}{dt} - \dot{\psi} \frac{d\delta t}{dt},$$

then one will find that:

$$\begin{aligned} \delta H \cdot dt &= dt \cdot \left[\sum \frac{\partial H}{\partial p_i} \delta p_i + \sum \frac{\partial H}{\partial \dot{p}_i} \delta \dot{p}_i \right] \\ &= dt \cdot \sum \frac{\partial H}{\partial p_i} \delta p_i + dt \cdot \sum \frac{\partial H}{\partial \dot{p}_i} \frac{d\delta p_i}{dt} - dt \cdot \sum \dot{p}_i \frac{\partial H}{\partial \dot{p}_i} \frac{d\delta t}{dt}, \end{aligned}$$

or when one recalls (II):

$$\delta H \cdot dt = dt \cdot \sum \frac{\partial H}{\partial p_i} \delta p_i + \frac{d}{dt} \left[\sum \frac{\partial H}{\partial \dot{p}_i} \delta p_i \right] dt - \sum \delta p_i \frac{d}{dt} \left(\frac{\partial H}{\partial \dot{p}_i} \right) - [E + H] \frac{d\delta t}{dt} dt$$

and

$$\int_{t_0}^{t_1} [\delta H \cdot dt + (H + E) \cdot d\delta t + \delta U \cdot dt] = 0 = \int_{t_0}^{t_1} dt \sum \delta p_i \left[\frac{\partial H}{\partial p_i} - \frac{d}{dt} \left(\frac{\partial H}{\partial \dot{p}_i} \right) + P_i \right] = 0,$$

since the term $\sum \frac{\partial H}{\partial \dot{p}_i} \delta p_i$ vanishes at the limits. That will then yield the generalized Lagrange equations in the form:

$$\frac{\partial H}{\partial p_i} - \frac{d}{dt} \left(\frac{\partial H}{\partial \dot{p}_i} \right) + P_i = 0.$$

If one multiplies those equations by $\dot{p}_1, \dots, \dot{p}_i, \dots$ in succession, adds them, and integrates over time then (as is known) one will find the equation for the conservation of energy:

$$dE = \sum P_i dp_i,$$

such that the E that is defined by (II) will actually represent the energy. One sees clearly from (II), as **Helmholtz** had already emphasized, that E is defined completely by H , but H cannot be determined uniquely from E . Namely, if F represents a *linear* function of \dot{p}_i , such that one has:

$$0 = \sum \dot{p}_i \frac{\partial F}{\partial \dot{p}_i} - F,$$

then one will also have:

$$E = \sum \dot{p}_i \frac{\partial (H + F)}{\partial \dot{p}_i} - (H + F),$$

such that H can then be increased by the arbitrary F , and *the same* E will nonetheless emerge from that.

Conversely, if the kinetic potential H is known for a reversible process then E can be calculated uniquely. It is only for purely-mechanical processes that one has $H = L - \Phi$, $E = L + \Phi$, in which Φ represents the potential energy.

Knowing the kinetic potential H as a function of p_i , \dot{p}_i , and the external work $\delta U = \sum P_i \delta p_i = \delta E$ makes it possible to apply the principle of least action even in those cases of reversible processes that are foreign to mechanics. If the variation of time is excluded, i.e., $d\delta t = 0$, then one will get back to the case that **Helmholtz** treated. Examples of that kind were found by **M. Planck** in the “Acht Vorlesungen über theoretische Physik,” pp. 100, *et seq.*

Here, all that shall be shown is how the equations of motion for an electron can be derived from the known kinetic potential H . One has:

$$H = -\mu c^2 \sqrt{1 - \frac{q^2}{c^2}} + \text{const.},$$

where $q^2 = \dot{x}^2 + \dot{y}^2 + \dot{z}^2$, μ is the mass for $q = 0$, and c is the speed of light. One then finds the kinetic energy:

$$E = \dot{x} \frac{\partial H}{\partial \dot{x}} + \dot{y} \frac{\partial H}{\partial \dot{y}} + \dot{z} \frac{\partial H}{\partial \dot{z}} - H = q \frac{\partial H}{\partial q} - H = \frac{\mu c^2}{\sqrt{1 - \frac{q^2}{c^2}}} + \text{const.},$$

and easily obtains from the rules above that:

$$\delta H \cdot dt + (H + E) d\delta t = \frac{\mu c^2}{\sqrt{1 - \frac{q^2}{c^2}}} \left[\dot{x} \frac{d\delta x}{dt} + \dots \right] = \frac{d}{dt} \left[\frac{\mu \dot{x} \delta x}{\sqrt{1 - \frac{q^2}{c^2}}} \right] - \delta x \frac{d}{dt} \left[\frac{\mu \dot{x}}{\sqrt{1 - \frac{q^2}{c^2}}} \right] + \dots$$

If one sets $\delta U = \mathfrak{F}_x \delta x + \mathfrak{F}_y \delta y + \mathfrak{F}_z \delta z$ and integrates over t in the sense of (I) then that will give:

$$\frac{d}{dt} \left(\frac{\mu \dot{x}}{\sqrt{1 - \frac{q^2}{c^2}}} \right) = \mathfrak{F}_x, \dots$$

In so doing, as is known:

$$\mathfrak{F}_x = e\mathfrak{E}_x + \frac{e}{c} [\dot{y} \mathfrak{H}_z - \dot{z} \mathfrak{H}_y], \dots,$$

in which \mathfrak{E} is the electric field strength, \mathfrak{H} is the magnetic field strength, and e represents the charge of the electron ⁽¹⁾.

⁽¹⁾ Cf., **Einstein**, Jahrb. der Radioaktivität (1907), pp. 433. – **Max Planck**, *loc. cit.*, pp. 123.